

**Demo PDF file. This file includes questions: 10 from 1890. Full version of file looks the same as demo, but full version includes all questions. You may download file with all questions by link on bottom of this page**

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## **MEWB - Steam Plants**

**1. Energy existing in a system as a result of the relative velocities of the objects is defined as \_\_\_\_\_.**

- potential energy
- pressure energy
- **kinetic energy**
- relative energy

Note:

*Kinetic energy is the energy possessed by an object due to its motion; therefore, it accurately describes energy resulting from the relative velocities of objects within a system. Potential energy relates to position, pressure energy to fluid mechanics, and 'relative energy' is not a recognized physics term.*

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**2. Which of the following statements is NOT one of Newton's laws?**

- A body at rest tends to remain rest and a body in motion tends to remain in motion.
- For every action there is an equal and opposite reaction.
- **If the pressure is constant, the volume of an enclosed dry gas varies directly with the absolute temperature.**
- An imbalance of force on a body tends to produce an acceleration in the direction of that force which is directly proportional to the applied force and inversely proportional to the mass of the body.

Note:

*The correct answer is choice C because it describes Charles's Law, a gas law relating volume and temperature, rather than a principle of motion or force as defined by Newton's laws.*

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**3. A theoretical engine cycle is a process that \_\_\_\_\_.**

- takes place in the combustor of the engine
- **begins with certain conditions, progresses through a series of additional conditions and returns to the original conditions**
- begins with certain conditions, progresses to a steady state and stays there
- None of the above.

Note:

*A theoretical engine cycle is defined as a sequence of processes that begins at a specific state and returns to that state, allowing for repetition.*

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**4. Boyle's law can best be defined as \_\_\_\_\_.**

- **the volume of an enclosed gas varies inversely with the applied pressure, provided the temperature remains constant**
- if the pressure is constant, the volume of an enclosed gas varies indirectly with absolute temperature
- a body at rest tends to remain at rest
- none of the above

Note:

*Boyle's law defines the inverse relationship between pressure and volume of a gas at constant temperature.*

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**5. Which of the following best describes the adiabatic pressure/volume relationship with regards to a fixed amount of a saturated gas?**

- **As pressure decreases, volume increases**
- As pressure increases, volume increases
- As pressure decreases, volume decreases
- As volume increases, pressure increases exponentially

Note:

*In an adiabatic process involving a fixed amount of saturated gas, pressure and volume are inversely related; therefore, as pressure decreases, volume increases.*

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**6. Which of the statements listed concerning heat transfer is correct?**

- Heat is always transferred at a constant rate.
- **Heat transfer rate increases as temperature difference increases.**
- The rate of heat transfer is not affected by temperature difference.
- The high temperature region is known as a heat sink.

Note:

*Heat transfer rate increases proportionally with temperature difference, as this difference drives heat flow. Heat transfer equations for conduction and convection demonstrate this relationship, while a heat sink is the lower temperature region, and heat transfer rates are rarely constant.*

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**7. A turbine exhaust pressure of 14.7 psia is equal to \_\_\_\_\_.**

- **00.000 psig**
- 01.470 psig
- 07.350 psig
- 29.400 psig

Note:

*14.7 psia represents atmospheric pressure, which is equivalent to 0 psig because psig measures pressure relative to atmospheric pressure.*

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**8. What is the gain in potential energy of an object weighing 10 pounds when it is raised to a height of 10 feet?**

- 10 ft-lb
- 20 ft-lb
- **100 ft-lb**
- 1000 ft-lb

Note:

*Potential energy gain is calculated by multiplying weight by height, resulting in 100 ft-lb (10 lb × 10 ft).*

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**9. How much work would be accomplished if a 15 pound box were raised to the top of a 25 foot platform?**

- 15 ft-lb
- 25 ft-lb
- 40 ft-lb
- **375 ft-lb**

Note:

*Work is calculated as force multiplied by distance; therefore, 15 pounds of force applied over 25 feet results in 375 ft-lb of work.*

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**10. What is the potential energy of an object weighing 10 pounds (4.54 kg) at a height of 10 feet (3.05 m)?**

- 10 ft-lb (1.38 kg-m)
- 20 ft-lb (2.76 kg-m)
- **100 ft-lb (13.83 kg-m)**
- 1000 ft-lb (138.25 kg-m)

Note:

*Potential energy is calculated by multiplying weight by height, resulting in 100 ft-lb (approximately 13.83 kg-m).*

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